

Extraction Chromatography Resins and their Use in Food Methods

Lawrence Jassin and Terence O'Brien

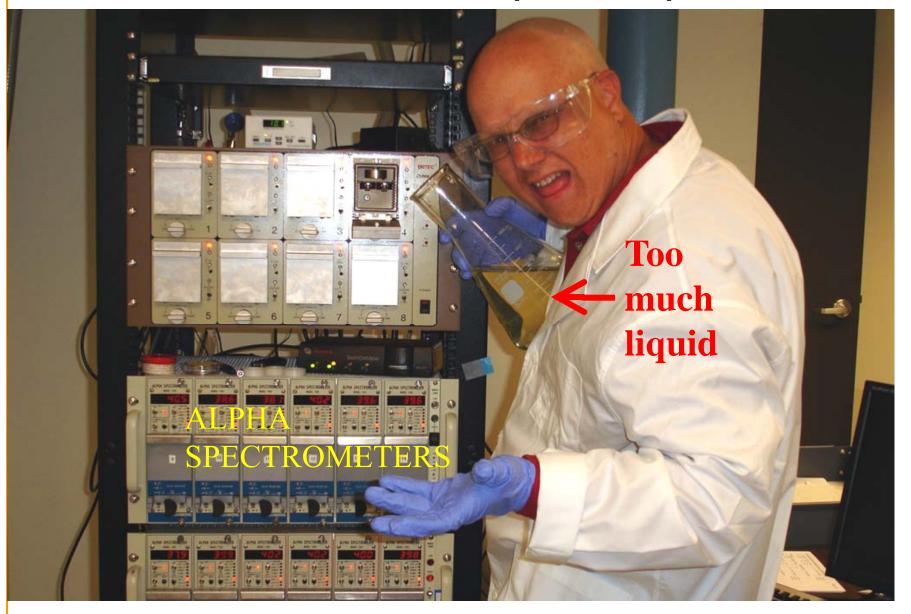
Winchester Engineering and Analytical Center (WEAC) July 15th, 2010

Variety is the spice of life



However, it is a **nightmare** in the Lab

eichrom We Need Sample Preparation



eichrom







Your separation resin drawer!



Hows

Versatility

Results

eichrom

Stops in the Lab/Kitchen along the way

- Why Sample Preparation / Instrumentation Choices
- Matrix Elimination Approaches
- Extraction Chromatography (EXC) Separations
- Some example separations and RESULTS
- Questions ?



Your goals for Sample Preparation

- Minimize sample prep time and waste generation
- Trade offs between
 - sample size, detection level, sample preparation rigor
- Data quality objectives
- Detection instrument selection
- Instrument's capabilities

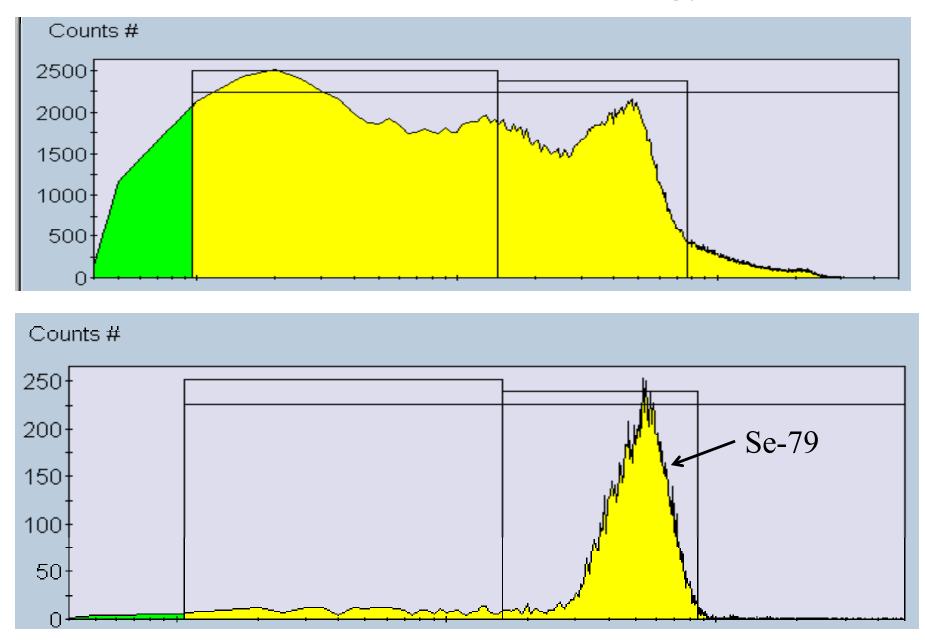
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Radiologic Screening Counters

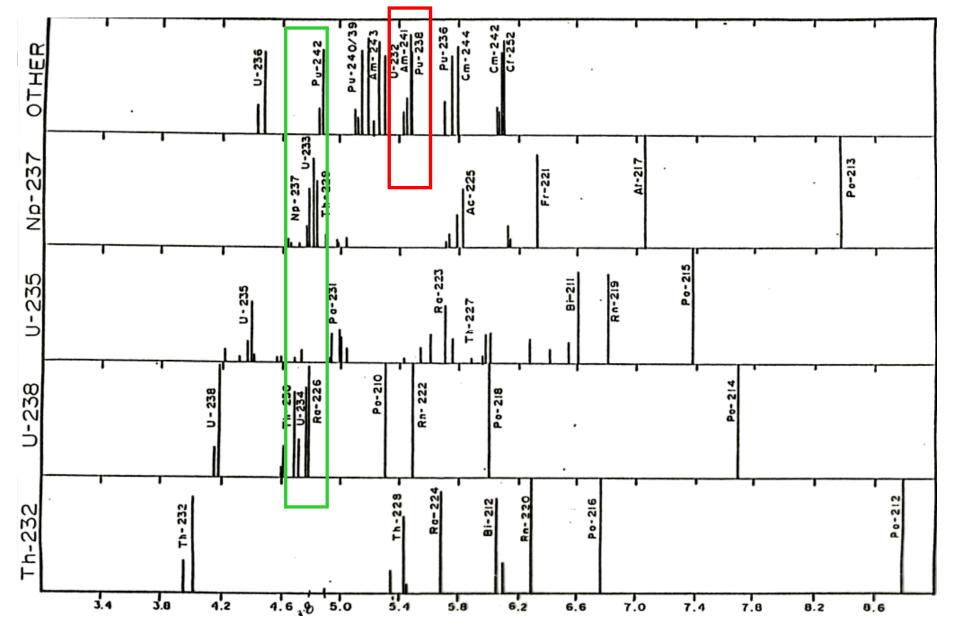
- Gas-Flow Proportional Counters
 - Alpha/Beta Weight Attenuation Curves
 - Alpha/Beta Cross Talk Calibration Curves
 - Alpha/Beta Isotope Calibration
- Liquid Scintillation Counters
 - Alpha/Beta Efficiency Determination
 - Alpha/Beta Cross Talk
 - Alpha/Beta Quench/Color Correction



eichrom Measurement of Beta Energy (LSC)



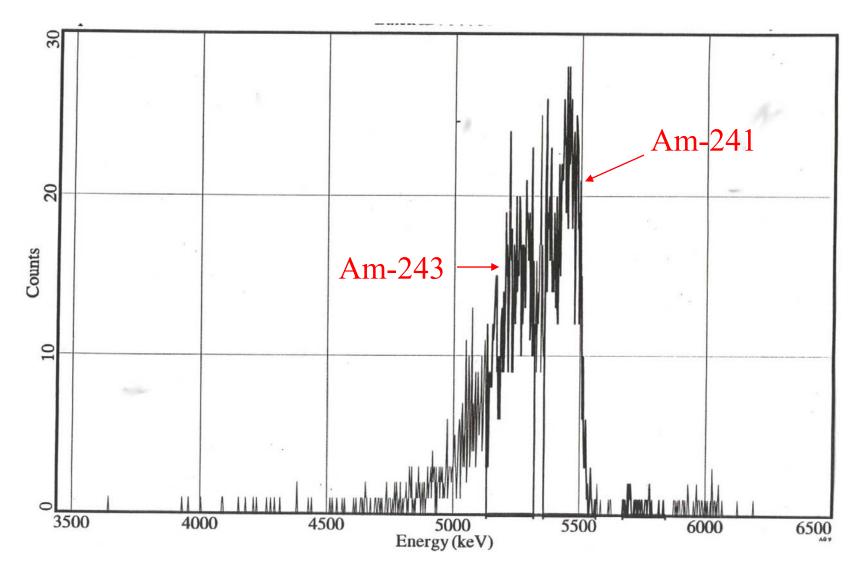
Measurement of Alpha Energy



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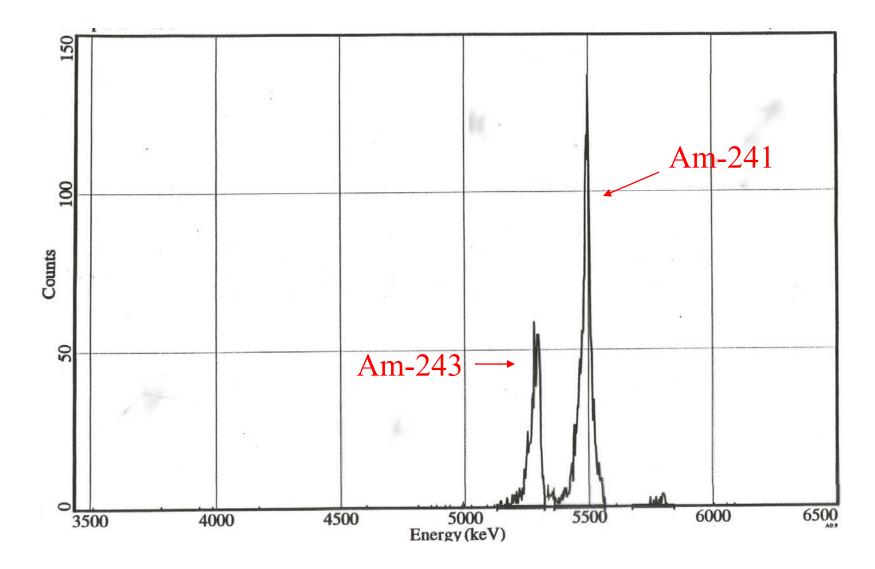
Americium Spectrum after TRU Resin Separation

presence of rare earths degrades spectrum- self absorption issues

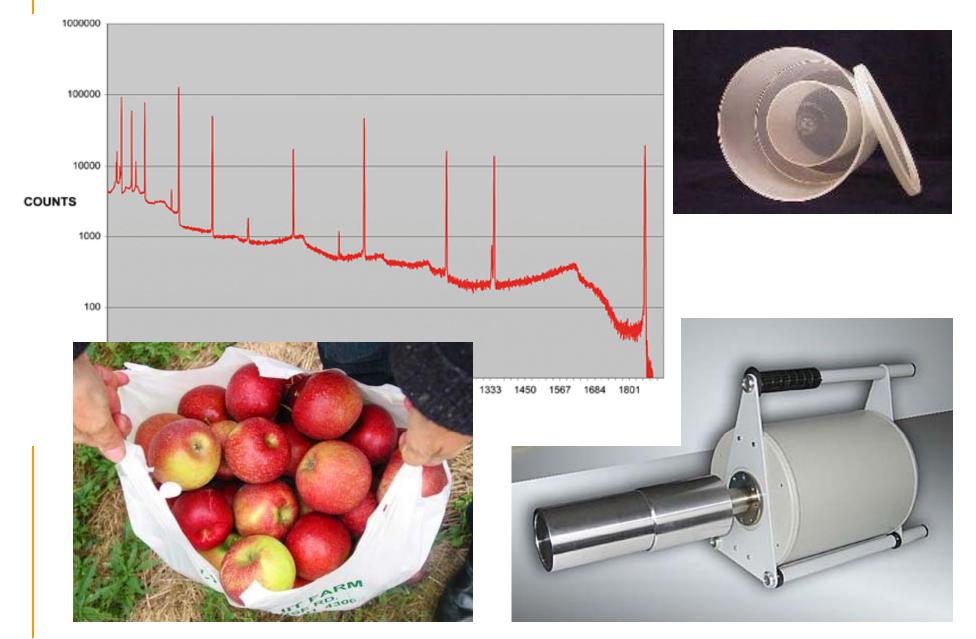


eichrom Am Spectrum after TEVA Resin Separation

Lanthanide elements removed - cleaner spectrum



eichrom Measurement of Gamma Energy



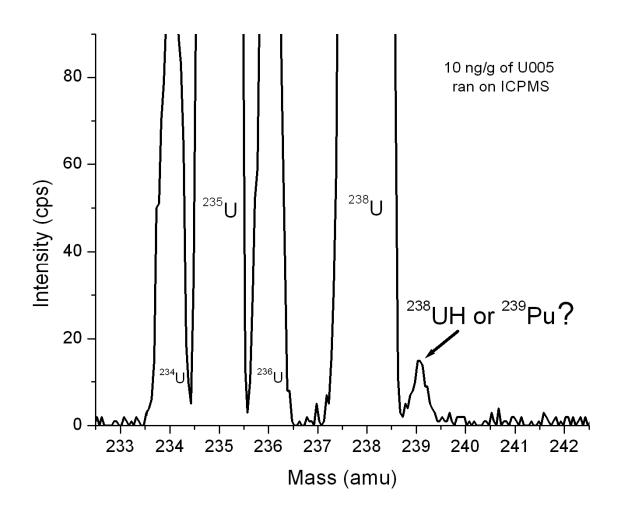
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Actinides by ICP-MS: The issues

Polyatomic ions

²³⁰ThH and ²³¹Pa
²³⁶UH and ²³⁷Np
²³⁷NpH and ²³⁸U
²³⁸UH and ²³⁹Pu
²⁴⁰PuH and ²⁴¹Am
²⁴²PuH and ²⁴³Am

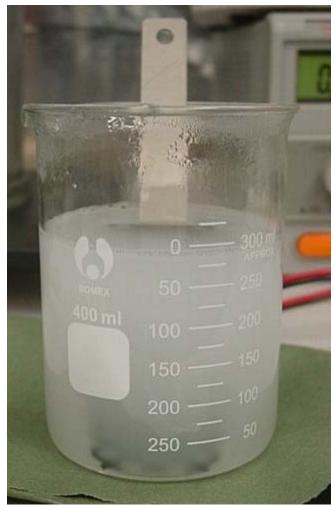
Isobars ²³⁶U and ²³⁶Np ²³⁸U and ²³⁸Pu ²⁴¹Pu and ²⁴¹Am ²⁴²Pu and ²⁴²Am





Matrix Elimination Approaches

- Sample Modification
 - Digestion
 - Leach
 - Fusion
- Volume Reduction
 - Precipitation
 - Evaporation
 - Ion Exchange



Actinides/Sr in Fish Method

- 200 g fish
- Wet ash
 - aqua regia/HNO₃/H₂O₂
- Furnace
 - 550°C



• Dissolve in 12 ml 6M HNO₃ +12 ml 2M Al(NO₃)₃ + 3M HNO₃ as needed (~40-45 ml load solution)

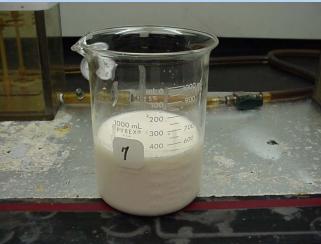


Sr in Milk

Calcium Phosphate Precipitation

- Milk Sample
 - 500 ml aliquot routine
 - 100 mL emergency + 50 mL Water to facilitate separation
- Add 2 mL 1.25M Ca (NO₃) $_2$ and 5 mL (NH₄) $_2$ HPO₄
 - Ca added so water blanks will precipitate (not really needed for milk)

500 ml milk



Calcium Phosphate Preciptation

- Add phenolphthalein indicator
- Add NH₄OH to dark pink (pH 10)
- Centrifuge 10 minutes at 3500 rpm
- Discard the supernatant

Acidification of Precipitant Separate the fat/protein from the Sr

- Add 20 mL of 3M HNO₃ into each tube
- Ca₃(PO₄) ₂ dissolves
- Fat/protein coagulate
- Centrifuge
- Transfer supernatant to beaker (contains the Sr)
- Rinse solids with 10-15 ml 3M HNO₃
- Transfer supernatant to beaker (remaining Sr)
- Evaporate beaker to dryness

Destruction of the remaining matrix

- Wet ash
 - 15 ml concentrated HNO₃ and 5 ml 30 wt% H_2O_2
- Heat beakers in a furnace
 - at 550C for 30-60 minutes to turn the solids white
- Wet ash
 - 10-15 ml concentrated HNO₃ and 5 ml 30 wt% H_2O_2
- Redissolve in 10 ml 8 M HNO₃ -1M Al(NO₃)₃

500 ml sample ppt. shown



$Ca_3 (PO_4)_2$ and fat/protein



Add 3M HNO3



Most fat/protein is removed

Centrifuge





Heat on hot plate





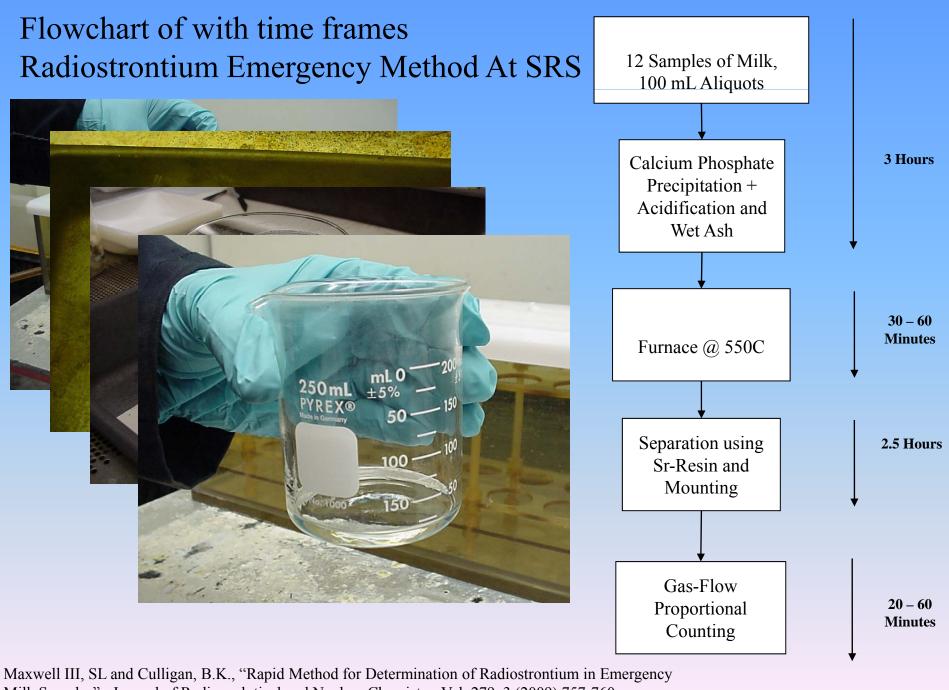
Heat to dryness





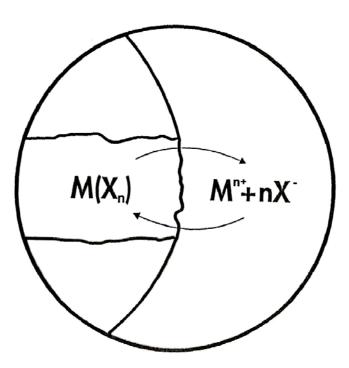
Heated at 550C 30 min.





Milk Samples", Journal of Radioanalytical and Nuclear Chemistry, Vol. 279, 3 (2009) 757-760

Metal Anion Complex Formation



Metal + Anion ____ Complex Complex + Organic ____ Extracted



Extraction Chromatography



Solvent Extraction

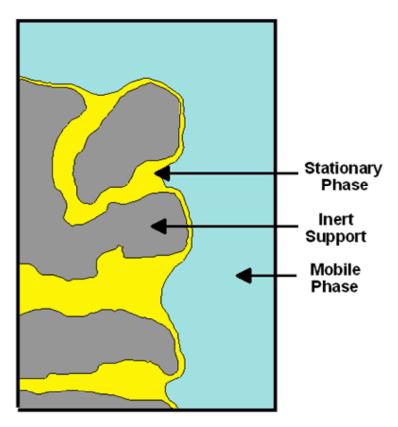


Column Chromatography



Extraction Chromatographic Resin

Surface of Porous Bead



Inert support =

Macroporous Acrylic Resin

Example Stationary Phases

•Crown Ether (Sr)

•CMPO (TRU)

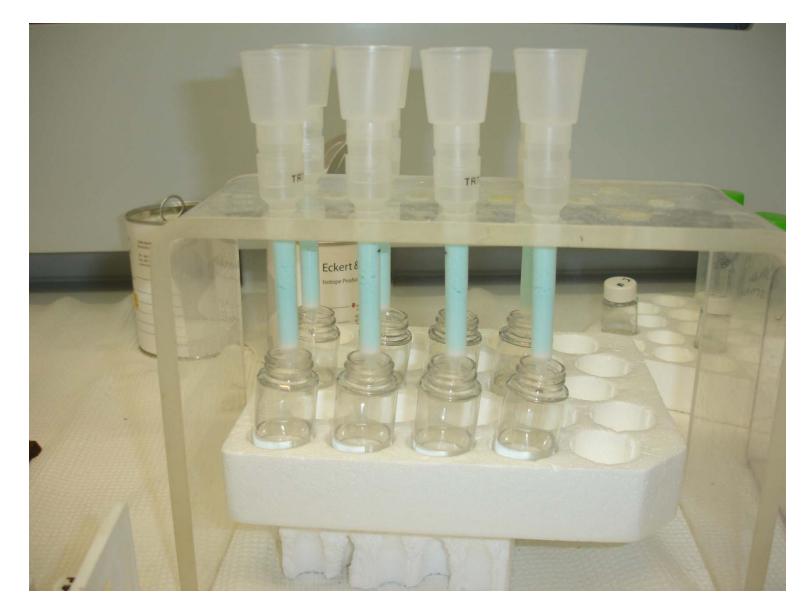
•DAAP (UTEVA)



eichrom Batch Reaction / MnO2 Resin for Ra Analysis







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Cartridge Format / Vacuum Assisted Flow

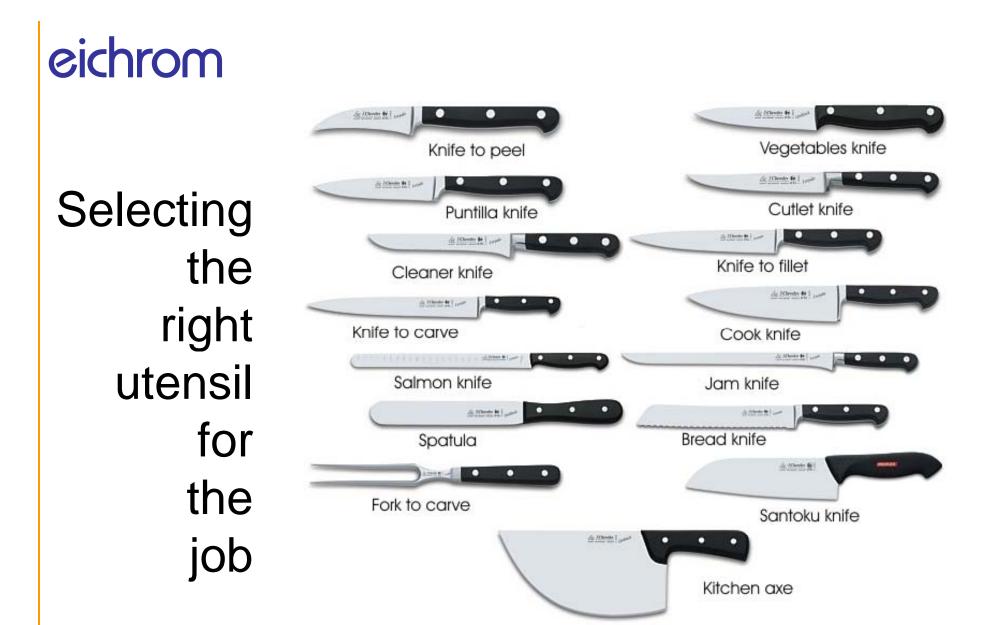


Gravity Flow vs. Vacuum Assisted



TEVA Discs/ Gravity or Vacuum Assisted Flow





expertise. collaboration. results.

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LATEST NEWS: Nuclear Power Outfitters Featured Product: "T-Flex" Tungsten Shielding System



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Product Name	Color Code	Applications*		
Accessories		Plastic accessories for columns and cartridges	product info	part numbers
Resolve™ Filters		Alpha spectroscopy source preparation	technical info	part numbers
Resolve™ PTFE Filters		NEW Radiological Air Monitoring	technical info	part numbers
Actinide Resin	• Yellow	Group actinide separations/gross alpha measurements	technical info	part numbers
Beryllium Resin		Ве	technical info	part numbers
DGA Resin		Actinids, Lanthanides, Y, Ra	technical info	part numbers
Diphonix® Resin		Actinides and transition metals	technical info	part numbers
lon Exchange Resins		Analytical grade cation and anion exchange resins	technical info	part numbers
Ln Resin	• Purple	Lanthanides, Ra-228	technical info	part numbers
MnO ₂ Resin		Ra	technical info	part numbers
Nickel Resin	• Pink	Ni	technical info	part numbers
Pb Resin	• Black	Pb	technical info	part numbers
Pre-filter Material		Organics removal	technical info	part numbers
RE Resin		Th, U, Np, Pu, Am, Cm, rare earth elements	technical info	part numbers
Sr Resin	• Red	Sr, Pb	technical info	part numbers
TEVA® Resin	• Green	Tc, Th, Np, Pu, Am/lanthanides	technical info	part numbers
Tritium Column		зН	technical info	part numbers
TRU Resin	• Blue	Fe, Th, Pa, U, Np, Pu, Am, Cm	technical info	part numbers
UTEVA® Resin	• Orange	Th, U, Np, Pu	technical info	part numbers

*Primary applications shown in blue.

EXC Resin Extractant Choice

- Ionic Recognition Extractant based resin

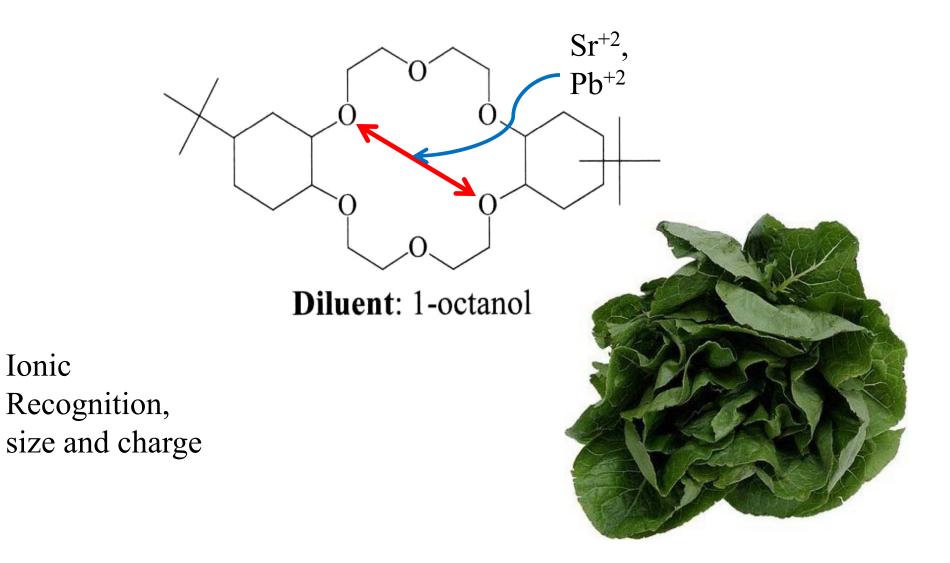
 Analyte retention related to charge and ionic radii
- Neutral and Anionic Extractant based resin
 - Analytes are directly hydrogen ion dependant, although anionic extractant based resins tend to show reduced analyte uptake as the acid competes
- Acidic Extractant based resin
 - Analyte retention is inverse hydrogen ion dependant

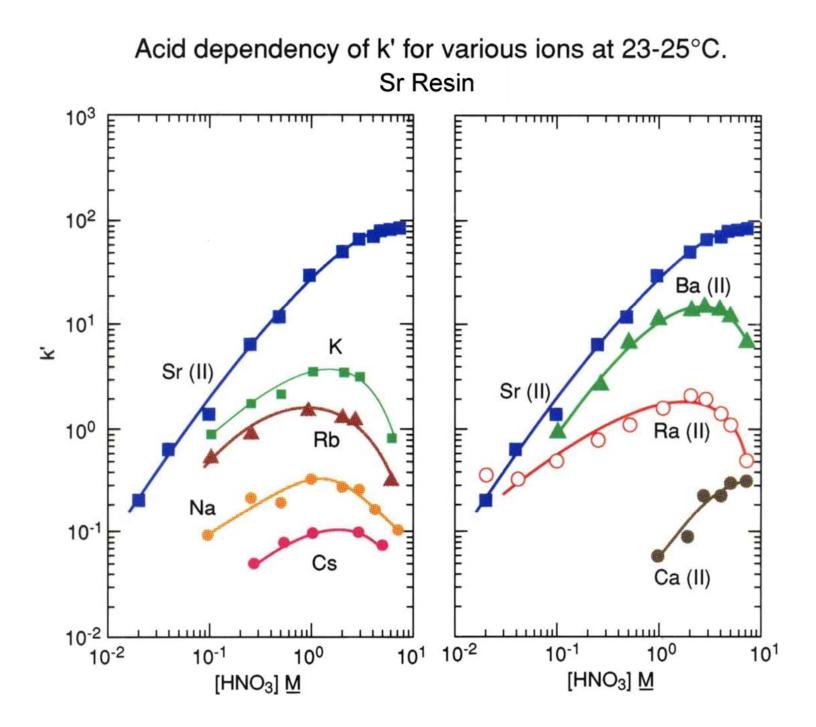


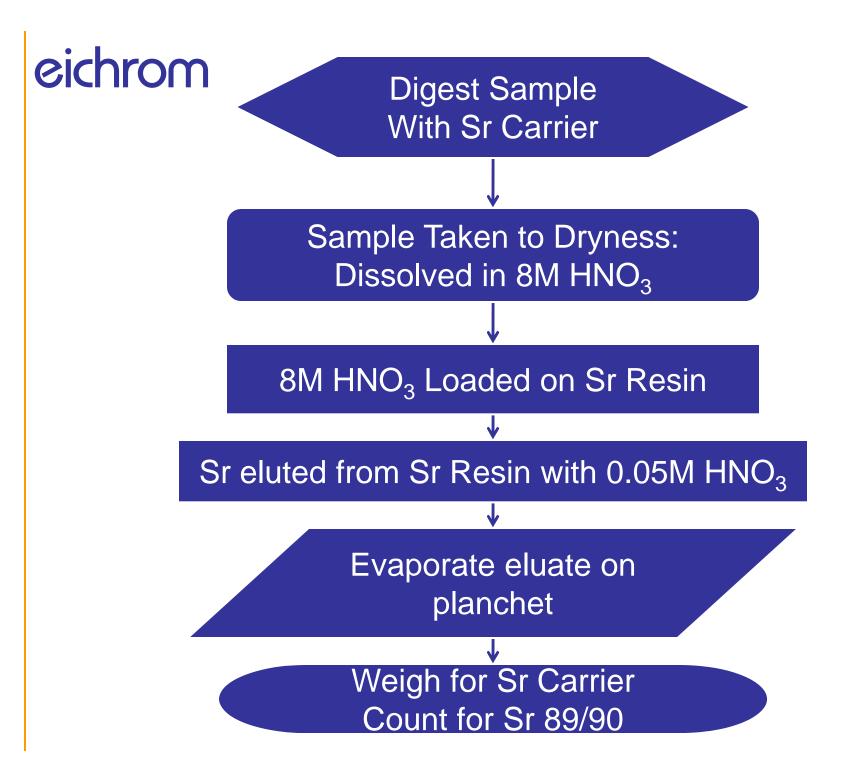
Ionic

Sr Resin

di-t-butylcyclohexano 18-crown-6







Sr-89/90 in Milk Column Extraction

- Redissolve in 10 ml 8M HNO₃-1M Al(NO₃) ₃
- Perform typical Sr Resin
 Separation using 3 ml Sr resin
 - (2 ml +1 ml cartridges)



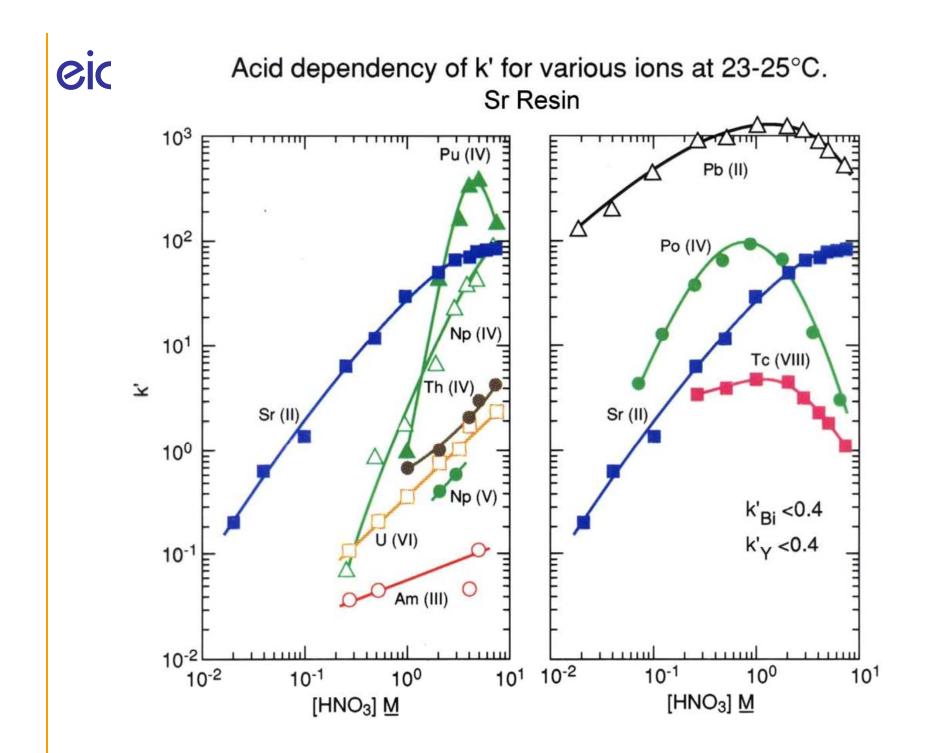
- Rinses:
 - \circ 15 mL of 8M HNO₃
 - \circ 5 ml 3M HNO₃-0.05M oxalic acid
 - \circ 7 ml 8M HNO₃
- Sr Elution: 13 ml 0.05M HNO₃

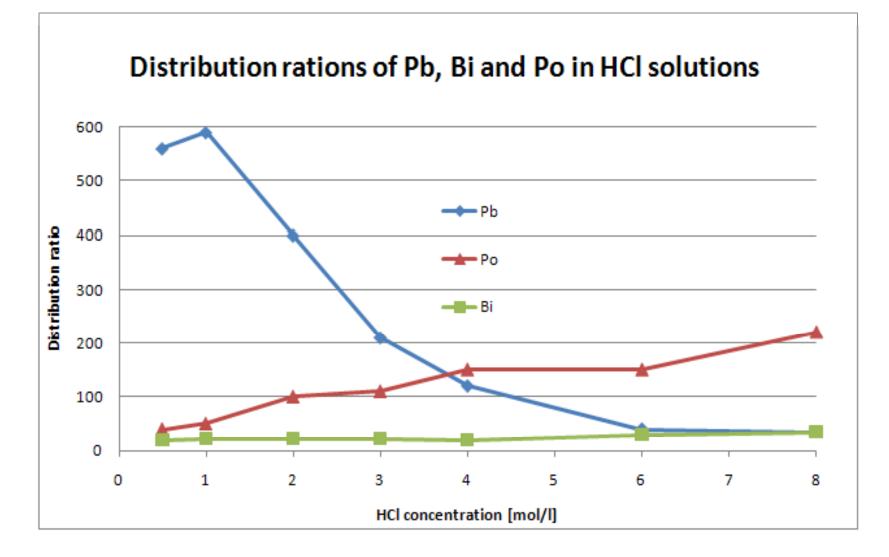
Performance of New Radiostrontium in Milk - 60 minute Count

⁹⁰ Sr Added	⁹⁰ Sr Measured	Uncertainty	Difference
(Bq/L)	(Bq/L)	(%, K=2)	(%)
0	0.11	130	N/A
0	0.27	59	N/A
2.86*	3.09	13.2	+8.0
2.86*	3.11	16.7	+8.7
2.86*	2.67	13.6	-6.6
2.86*	2.67	11.3	-6.6
5.70	5.85	10.4	+2.6
5.70	5.75	8.3	+0.9
5.70	6.04	8.2	+5.9
14.3	13.6	6.1	-4.9
14.3	14.0	6.1	-2.1
14.3	14.2	6.1	-0.7

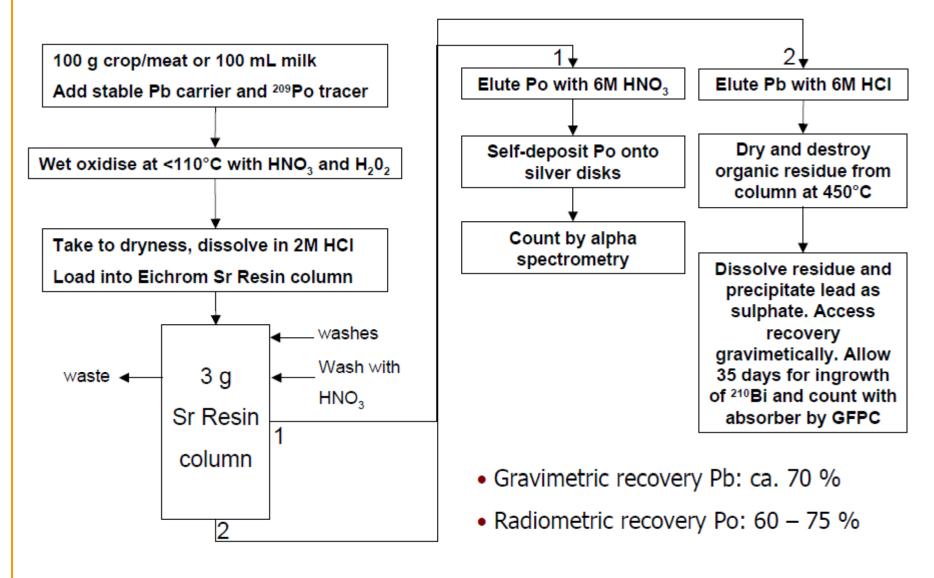
* Added using NRIP water standard

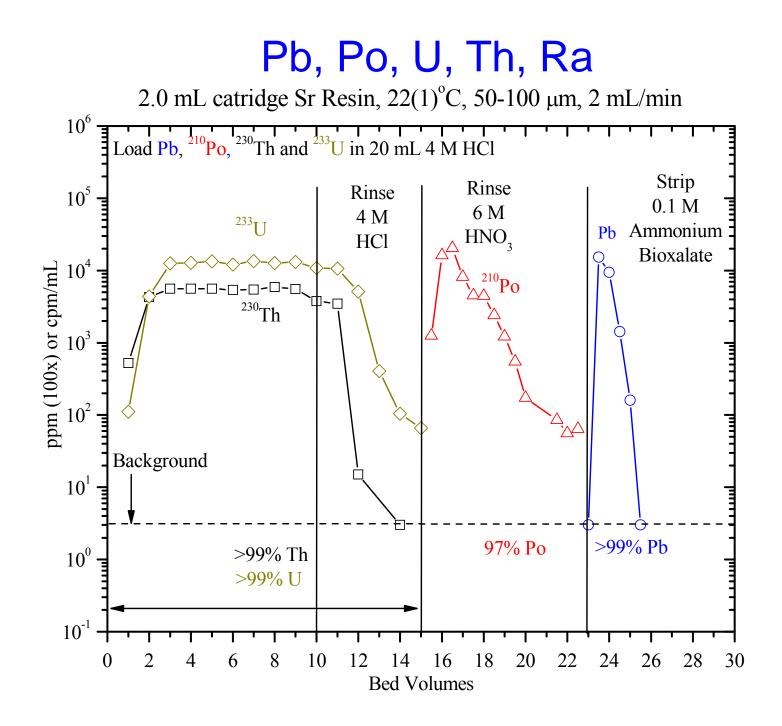
Average +0.52





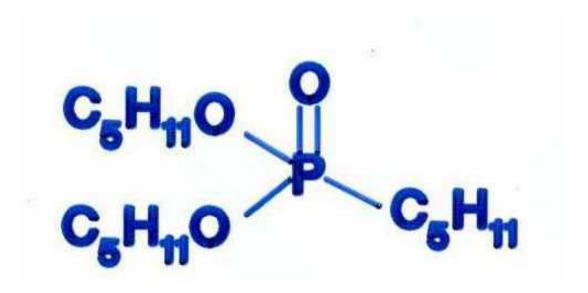
Pb and Po in milk, crop and animal samples (Dell)

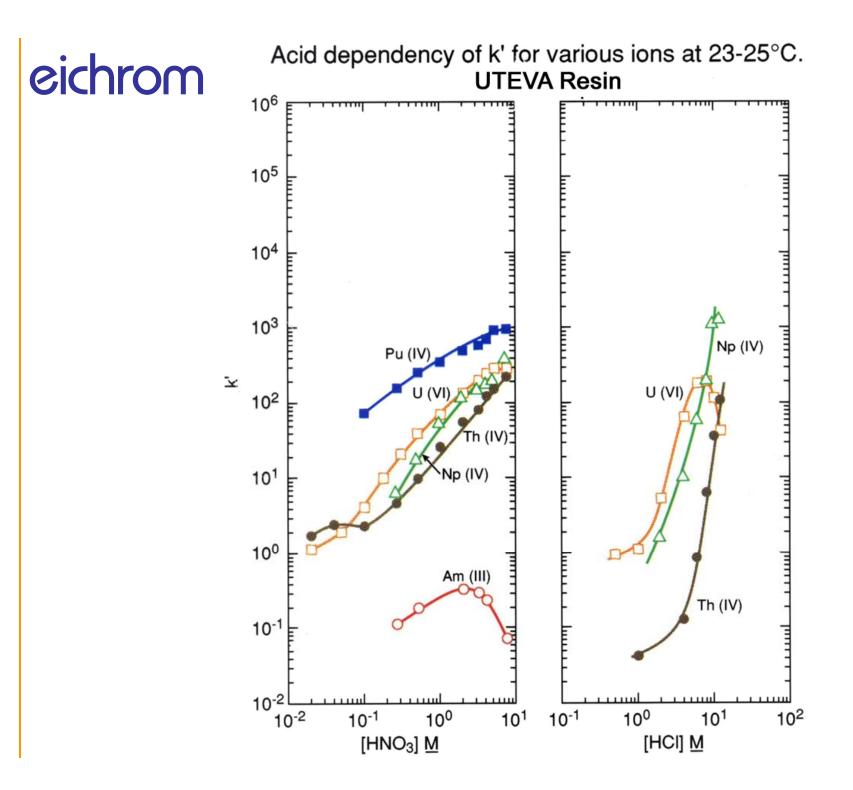






UTEVA Resin / Neutral Extractant Diamyl Amylphosphonate (DAAP) a.k.a. Dipentyl Pentylphosphonate (DPPP)





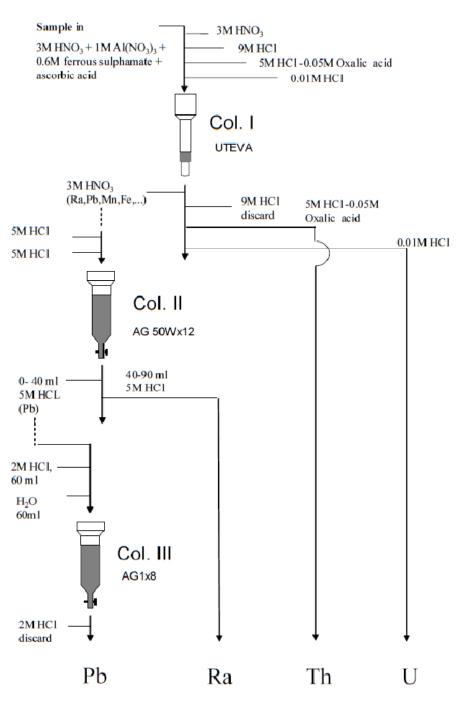
UTEVA

Sequential extraction procedure for determination of U, Th, Ra, Pb and Po radionuclides by alpha spectrometry in Environmental Samples

Isotopic <u>Tracer</u> Recovery Yields (mean $\pm 1\sigma$) for 19 Biota Samples

Analyte	Yield	Uncertainty
U	0.88	±0.12
Th	0.47	±0.18
Ra	0.22	±0.12
Po	0.80	±0.08
Pb	0.28	±0.24

CJ. M. Oliveira and F. P. Carvalho



Absorption of elements from HCl solutions by Anion Resin

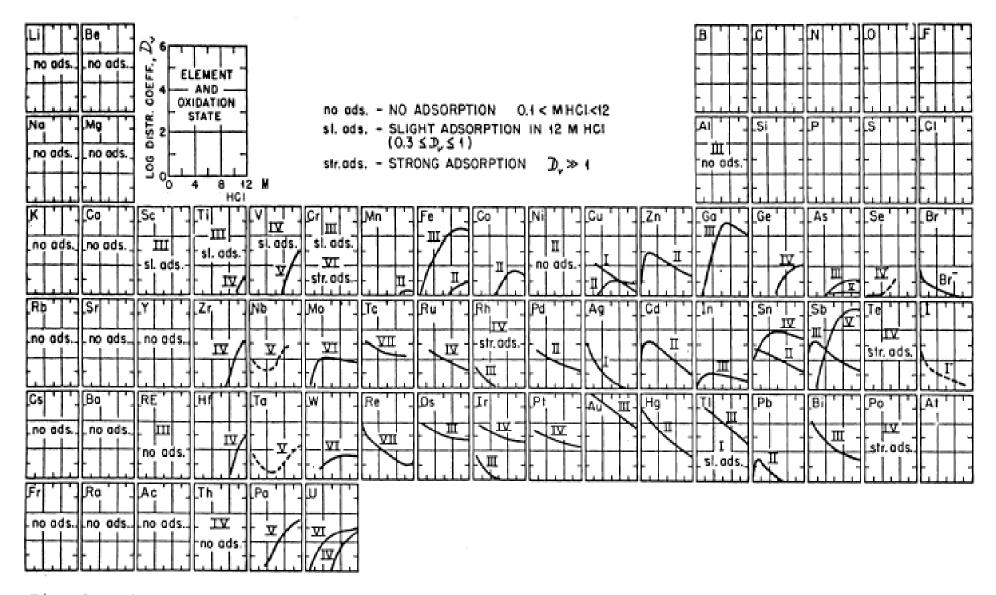
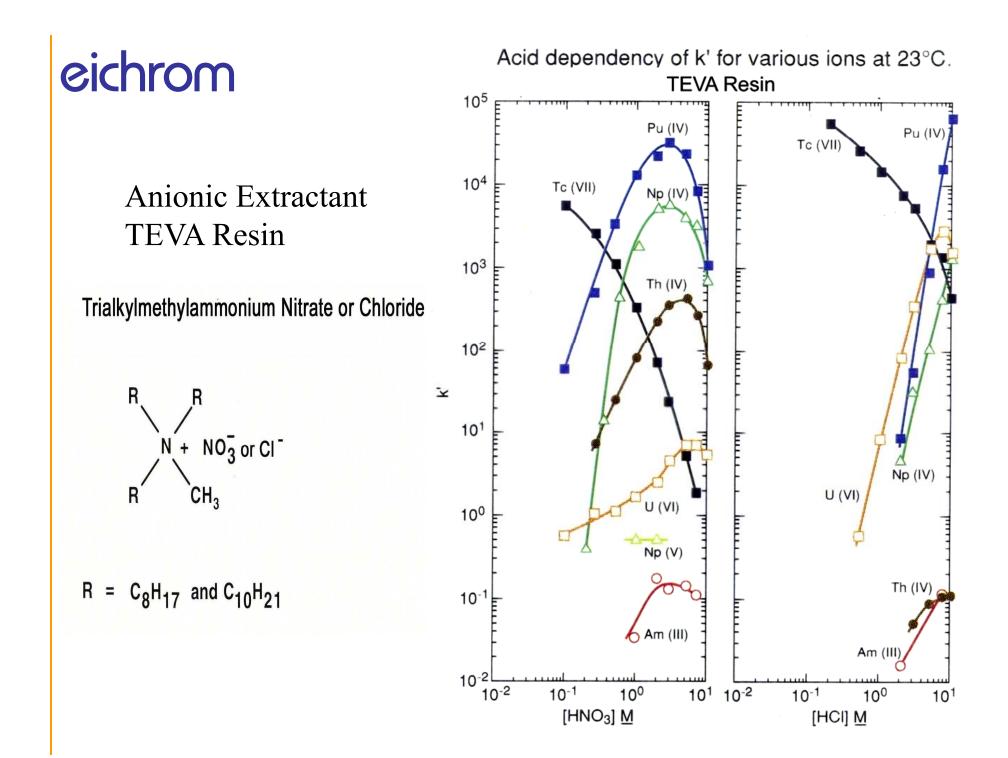


Fig. 6. Anion exchange distribution coefficients in HCl solutions.(Dowex 1-X10) (Ref. 3)

Li le de		в С	NO		
		I			
NO ADS. NO ADS.					
	9 2				
		III	x		
NO ADS. NO ADS. S OUT 5 10 14 M	NO ADS NO ADSORPTION FROM 0.1-14 M HNO3	NO ADS.	NO ADS.		
	SL, ADS SLIGHT ADSORPTION				
K Co Sc Ti V Cr	Mn Fe Co Ni Cu Zn	Go Ge	As Se		
		π	X		
NO ADS. NO ADS. NO ADS. NO ADS. NO ADS. NO ADS.	المتعادي والمتعادين والمتعادية والمتعادية والمتعادية والمتعادية والمتعادية والمتعادية والمتعادية والمتعادية والمتعادية	NO ADS.	NO ADS.		
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Rb Sr Y Zr Nb Mo	Tc Ru Rh Pd Ag Cd	In Sn	Sb Te		
THE TRANSPORT		п	X IX		
NO ADS. NO ADS. NO ADS. SL. ADS.	SL. ADS. NO ADS. NO ADS.	NO ADS. SL. ADS	NO ADS. NO ADS.		
	Re Os Ir Pi Au Hg	TI P6	Bi Po		
		Τ	π		
NO ADS. NO ADS. SL. ADS.	NO ADS.	SL ADS.			
		I			
Fr Ro					
NQ ADS.					
Lo Ce Pr Nd Pm Sm	Eu Gd To Dy Ho Er	Tm Yb			
		п п	H		
SL. ADS. NO ADS. NO ADS.		NO ADS. NO ADS			
AC Th Po U No Pu		Md No			
NO ADS	NO ADS. NO AOS.				
T T NO ADS					
Fig. 7. Anion exchange distribution coefficients in HNO2 solutions. (Dowex 1-X10)					
Fig. 7. Anion exchange distribution	ition coefficients in HNO2 solution	ons.(Dowex	1 - X10		

Absorption of elements from HNO₃ solutions by Anion Resin

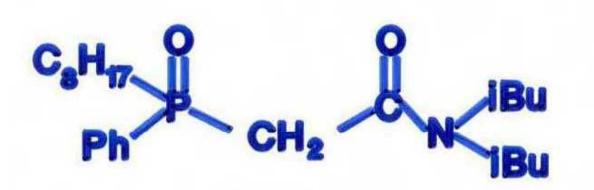
Fig. 7. Anion exchange distribution coefficients in HNO3 solutions.(Dowex 1-X10) (Ref. 4)



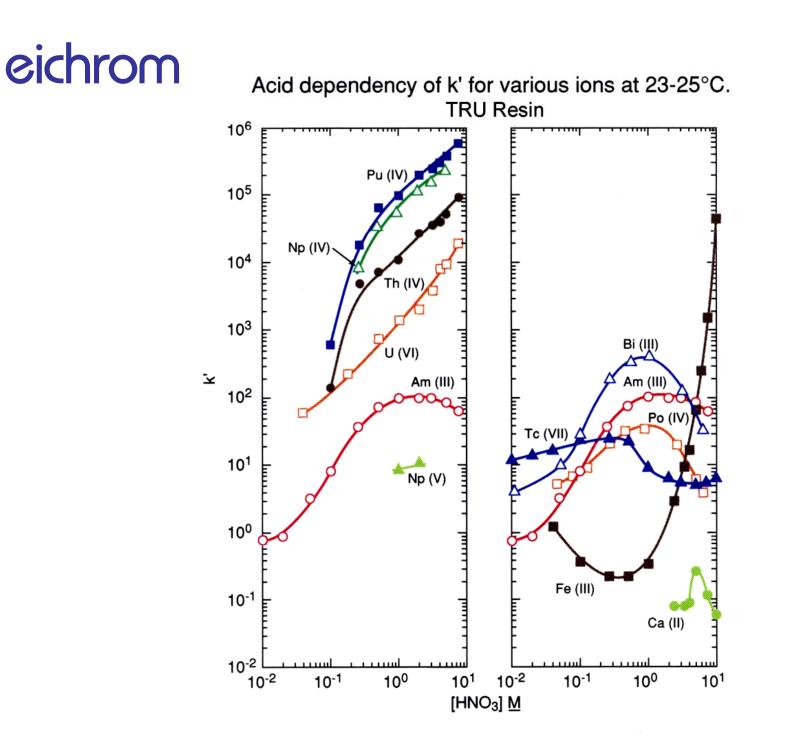


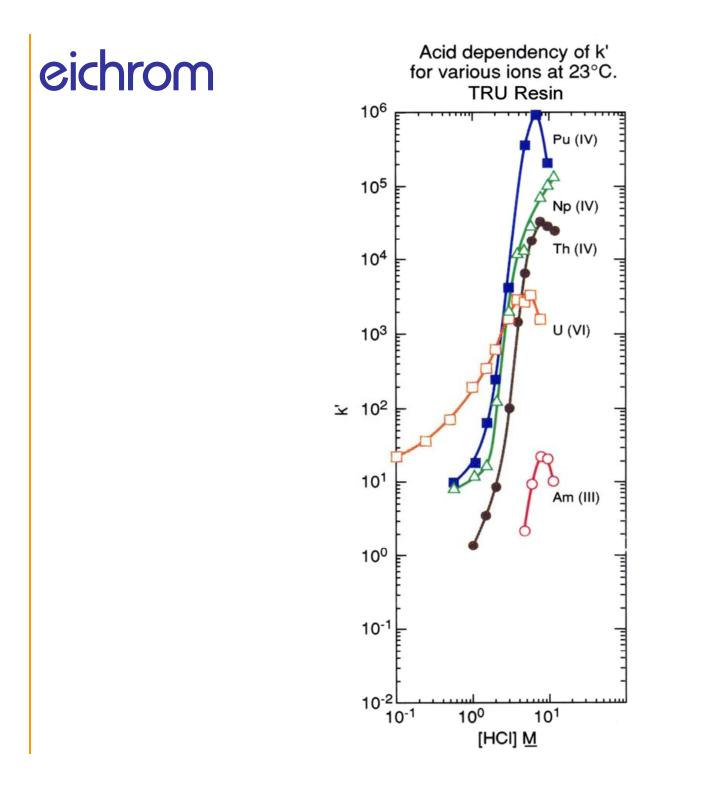
Neutral Extractant - TRU Resin, Octyl(phenyl)-N,N-diisobutylcarbamoylmethylphosphine oxide

(CMPO)



Diluent: tri-*n*-butyl phosphate





Actinides/Sr in Fish Method



- Load to TEVA+TRU+DGA
 - after valence adjustment
 - using sulfamic acid, iron (if Np-237 needed), ascorbic acid, followed by sodium nitrite
- Collect load/rinse (evaporate and redissolve later in 8M HNO3 for Sr Resin)

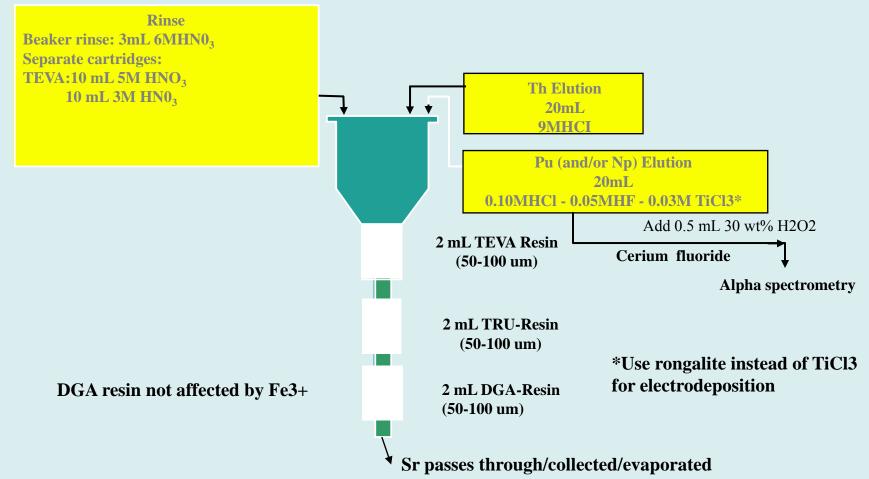


Actinides and Sr-90 in Fish Data

Tracer/carrier Recovery	Avg. Recovery	MS
Pu-236	99.8%	100% (Pu-238) 90.0% (Np-237)
Am-243	109%	94.1% (Am-241) 94.3% (Cm-244)
U-232	97.1%	91.1% (U-235)
Sr carrier	84.9%	97.7% (Sr-90)

Actinide Column Separation

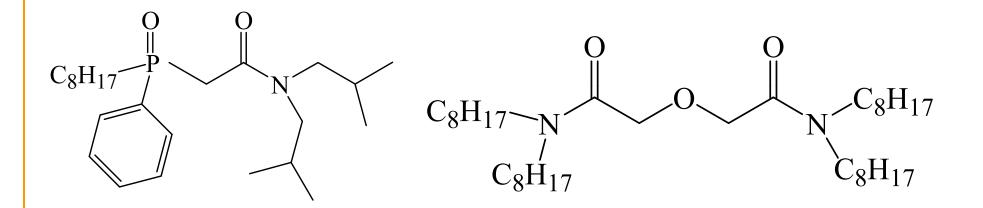
- 1) Redissolve in 7 mL warm 6M HN0₃ and 7mL 2M AI(N0₃) $_3$
- Add 0.5 mL 1.5M Sulfamic Acid + 1.25 mL 1.5M Ascorbic Acid/ 1 mg Fe (if Np-237 analyzed)
- 3) Add 1 mL 3.5 M Sodium Nitrite





TRU Resin: Neutral Extractant

DGA Resin: Neutral extractant/ ionic recognition

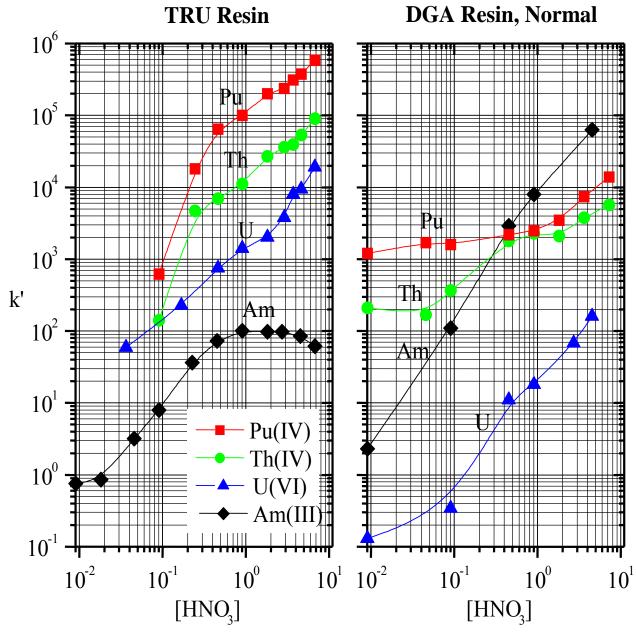


TRU (CMPO)

DGA

 $Am^{3+} + 3X^{-} + \overline{3E} \leftrightarrow \overline{AmX_3E_3}$ $X = Cl^{-} \text{ or } NO_3^{-}$

Actinides on TRU vs. DGA out of HNO₃





- DGA alone:
 - Rinse DGA with 8 ml 0.1M HNO3
 - Add rinse to initial load/rinse solution containing Sr and evaporate
 - Contains some Sr
 - Place TRU cartridge above DGA and elute Am/Cm from TRU onto DGA with 15 ml 4M HCL
 - Discard rinse
- Rinse DGA (alone) with 3 ml 1M HNO3/10 ml 0.1M HNO3 to remove interferences
 - Discard rinse
- Elute Am/Cm from DGA with 10 ml 0.25M HCl
 - Cerium fluoride microprecipitation
 - Alpha spectrometry

eichrom Actinides/Sr in Fish Method

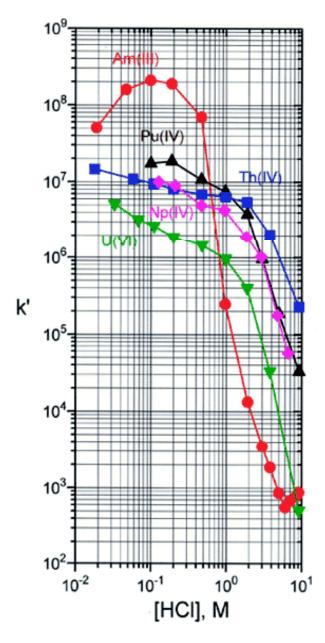
- Rinse TRU with 12 ml 4M HCL-0.2M HF
 - Th removal
- Elute U from TRU with 15 ml 0.1M ammonium bioxalate
 - Cerium fluoride precipitation/alpha spectrometry



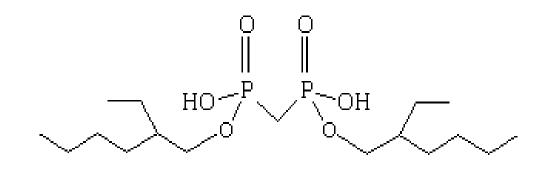
Typical Actinide Tracers and Sr Carrier Recoveries for Animal Tissue Matrices

Matrix	Pu-236	Am-243	<u>U-232</u>	Sr Carrier
Beef (N=6)	98.7% ±5.7%	97.1% ± 8.4%	93.4% ±4.7%	96.3% ±0.5%
Deer (N=59)	99.3% ±12%	93.4% ±10%	90.4% ±8.0%	83.4%±3.5%
Fish-Bass (N=72)	96.2% ±14%	101.8%±13%	95.1% ±8.1%	89.0% ±16%
Fish-Bream (N=57)	96.6% ±12%	98.4% ±7.7%	91.1% ±6.3%	91.7% ±10%
Fish-Catfish (N=69)	98.3% ±12%	103.7% ±7.6%	89.4% ±12%	89.4% ±17%
Fish-Mullet (N=6)	96.2% ±6.8%	100.4% ±8.9%	91.0% ±8.1%	85.6% ±17%
Fish-Red Fish (N=6)	99.5% ±11%	105.2% ±8.6%	95.7% ±3.2%	77.7% ±21%
Fish-Sea Trout (N=6)	100.5% ±5.0%	102.2% ±7.6%	83.5% ±20%	74.4% ±25%
Hog (N=17)	93.0% ±20%	96.4% ±9.7%	86.4% ±15%	86.0%±7.1%
Shellfish (N=5)	101.3 ±2.2%	97.4% ±7.1%	81.7 ±3.2%	97.5% ±0.89%

Composite tissue samples 100 gram-deer, hog, bream, shellfish 200 gram-catfish, bass, red drum, mullet, sea trout 25 gram- nonedible fish samples including bones



Actinide Resin uptake of various actinides with DIPEX® extractant (Liquid Chelating Exchanger)



Data developed at Argonne National Laboratory, USA

eichrom Your separation resin drawer!



Hows Versatility

Results



¿ Questions ?

